

MTBE-ONLY GROUNDWATER RELEASES

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**SAM Forum & Symposium on MTBE and
other Oxygenates**

September 25, 2002
San Diego, CA

CONTAINS

MTBE

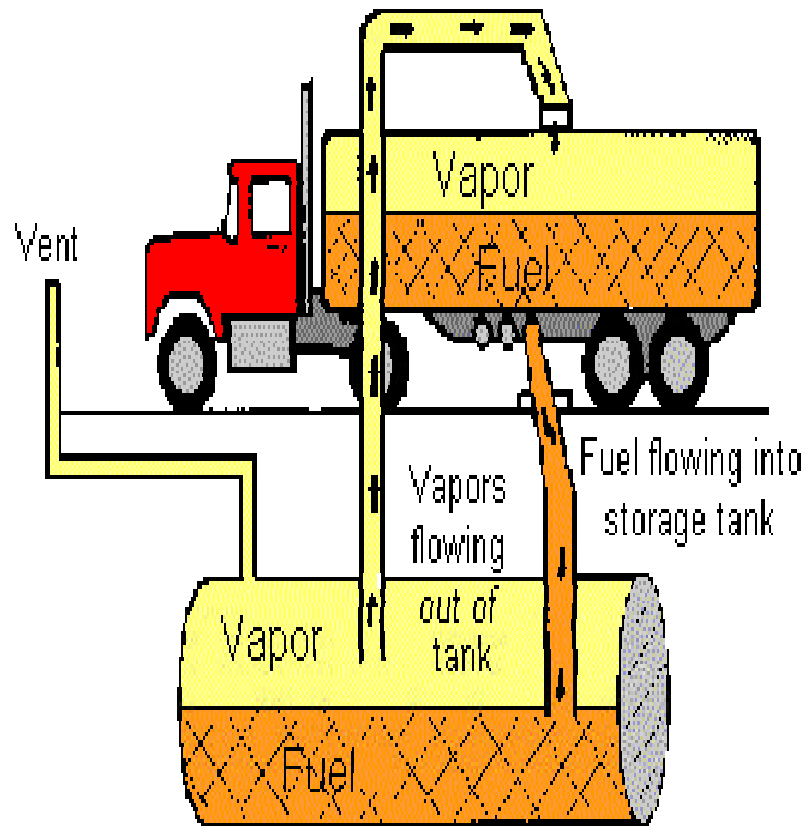
THE STATE OF CALIFORNIA HAS DETERMINED
THAT THE USE OF THIS CHEMICAL PRESENTS
A SIGNIFICANT RISK TO THE ENVIRONMENT

UST Release Scenarios

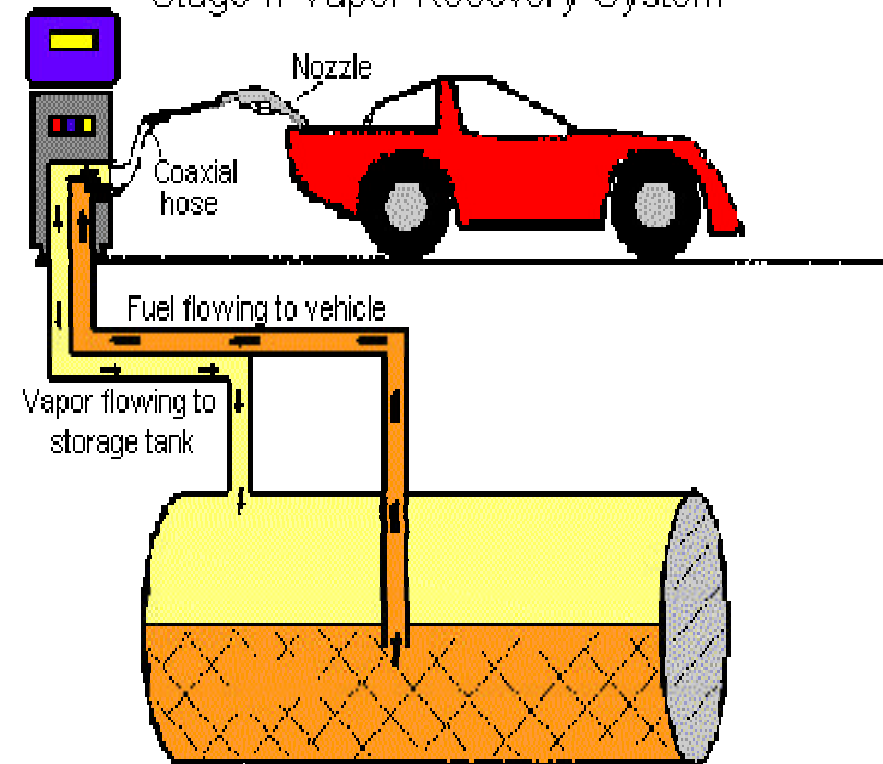
- **Catastrophic Release of Gasoline**
- **Gasoline Release**
w/ retained PSH (“soils-only” release)
- **Vapor**
- **All of the above**

UST Vapor Systems

Stage I Vapor Recovery System



Stage II Vapor Recovery System



Vapor Problems?

Underground Storage Tank System Field-Based Research Project Report,
submitted to SWRCB May 31, 2002

- 182 UST upgraded Systems tested
- 61% had detectable vapor leaks
- Majority of leaks in tank excavation
- Many detected at tank top (fill/vent risers)
- None likely exceeded UST integrity test standard of 0.1 gph (2.4 gallons/day....)
- Hear more from SWRCB at 1:40 today!

The Great Escape, 1

LUSTLine Bulletin 30, by Blayne Hartman (www.tegenv.com)

What vapors are coming out of the UST?

First...Calculate vapor pressures of gasoline compounds

$$P_i = VP_i * MF_i$$

VP_i , is the Vapor Pressure of component i

MF_i , is the Mole Fraction of component i

The Great Escape, 2

Then...Convert Pressures into relative concentrations

$$C_i = VP_i * MW_i * MF_i / RT$$

C_i , concentration (mass/liter) of component i

MW_i , molecular weight of component i

MF_i , is the mole fraction of component i

RT , is the universal gas constant times
temperature

The Great Escape...of MTBE

	VP (atm)	MW (g/M)	MF	C _{vapor} (ug/L)
Benzene	0.13	78	0.025	<u>10,600</u>
MTBE	0.32	88	0.125	<u>147,000</u>
Alkanes (C4-C8)	0.2	100	0.50	400,000

Chronic Release

“soils-only” w/ PSH

- Residual gasoline in vadose zone
- Soils are now the “UST”
- Calculation is the same as for the UST
- Apply Equation 6-13 in SAM Manual and in vapor risk spreadsheet
(for TPH >100 mg/kg)

MTBE Vapor and Water

Ref: SAM Manual

Multiphase Equilibrium

$$C_{\text{total}} = C_{\text{vapor}} + C_{\text{water}} + C_{\text{soil}} (\dots + C_{\text{PSH}})$$

.....+ PSH as condensate...?

$$C_{\text{vapor}} = H C_{\text{water}}$$

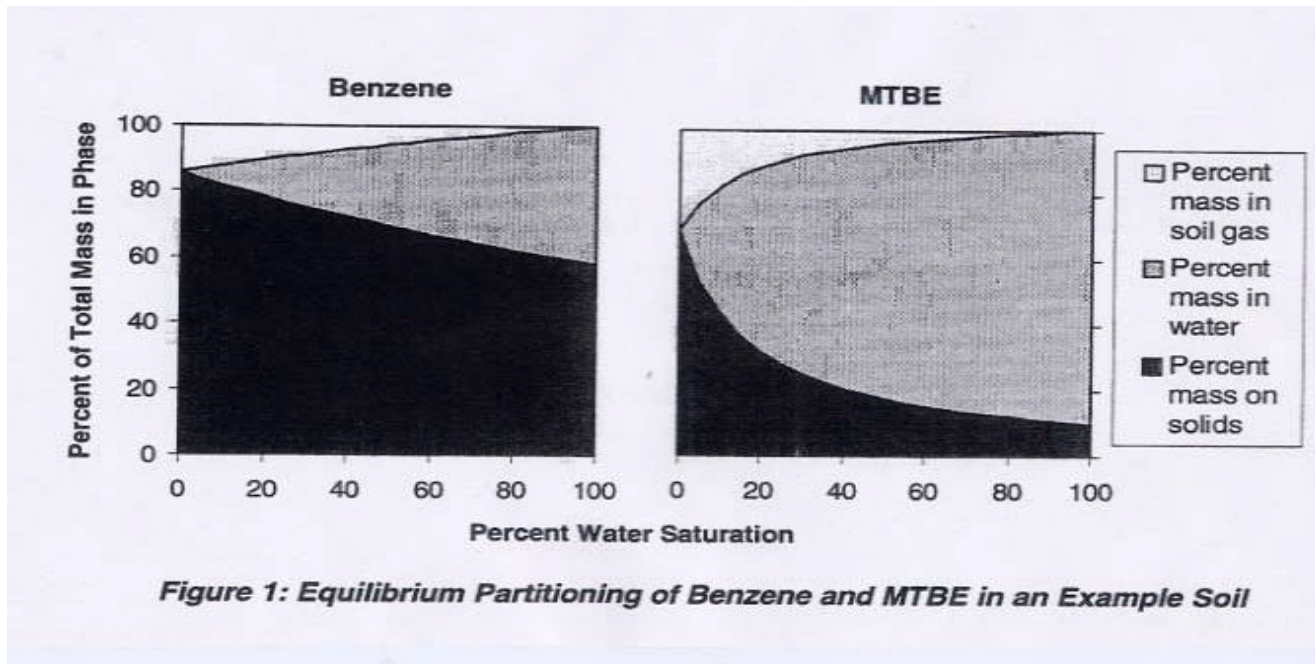
$$C_{\text{soil}} = K_{\text{oc}} F_{\text{oc}} C_{\text{w}}$$

	H, (d'less)	S, mg/L	K _{oc}
Benzene	0.23	1800	62
MTBE	0.024	48,000	78

“Example Soil”

Strategies for Characterizing subsurface Releases of Gasoline containing MTBE, API, August 2000, No. 11

- Calculated equilibrium distribution of Benzene and MTBE in an unsaturated soil
- Note: persistence of MTBE vapor



Vapor Movement, 1

Blayne Hartman, LUSTLine Bulletin 28/ SAM Manual

- In absence of recharge, vapors will diffuse
- Effective Diffusivity

$$D_e = D_{\text{air}} (\Theta_{\text{air}}^{10/3} / \Theta_{\text{total}}^2) \quad (\text{SAM Manual})$$

For MTBE, SAM default soil

$$\Theta_{\text{air}} = 0.20$$

$$\Theta_{\text{total}} = 0.30$$

$$D_{\text{air}} = 0.08$$

$$D_e = 0.00418 \text{ cm}^2/\text{sec}$$

Vapor Movement, 2

Blayne Hartman, LUSTLine Bulletin 28

- Diffusive velocity: no advection or recharge
- Distance over time (w/continuous source strength)

$$\text{Distance} = (2 * D_e * \text{time})^{1/2}$$

$D_e = 0.00418 \text{ cm}^2/\text{sec}$ for MTBE; time = 1 year

Distance = 513 cm (17 feet) per year

.....Direct Implications for shallow groundwater
MTBE-only release from vapor phase

Vadose Zone Transport

Simulation of Transport of MTBE to Groundwater From Small-Volume Releases of Gasoline in the Vadose Zone.

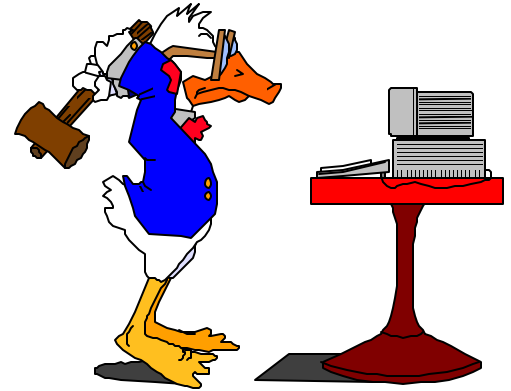
Lahvis and Rehmann API Research Bulletin, June 2000, no. 10

- Based on USGS Model R-UNSAT
by Matthew Lahvis and Arthur Baehr
- Multi-species transport
- Variable moisture content
- Diffusion, recharge, sorption
- O₂ and biodegradation

Thanks to Dr. Matt Lahvis, Shell Global Solutions (US) Inc. for following graphics

WHAT DO WE KNOW APRIORI (MTBE vs. C_6H_6)

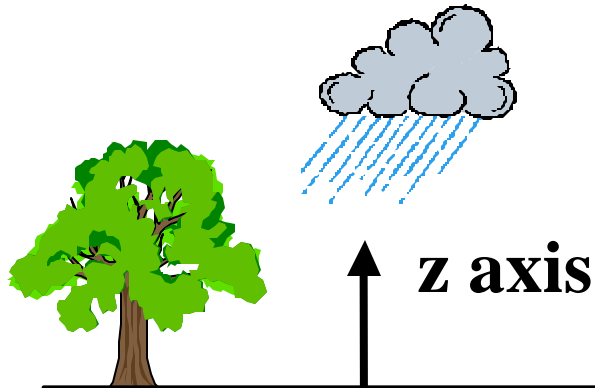
- source composition
- phase-partitioning properties
- principal transport mechanisms



ASSUMPTIONS

- **source is immobile**
- **MTBE is non reactive, benzene is reactive**
- **hydrostatic moisture distribution**
- **soil properties (van Genuchten)**
- **equilibrium partitioning**

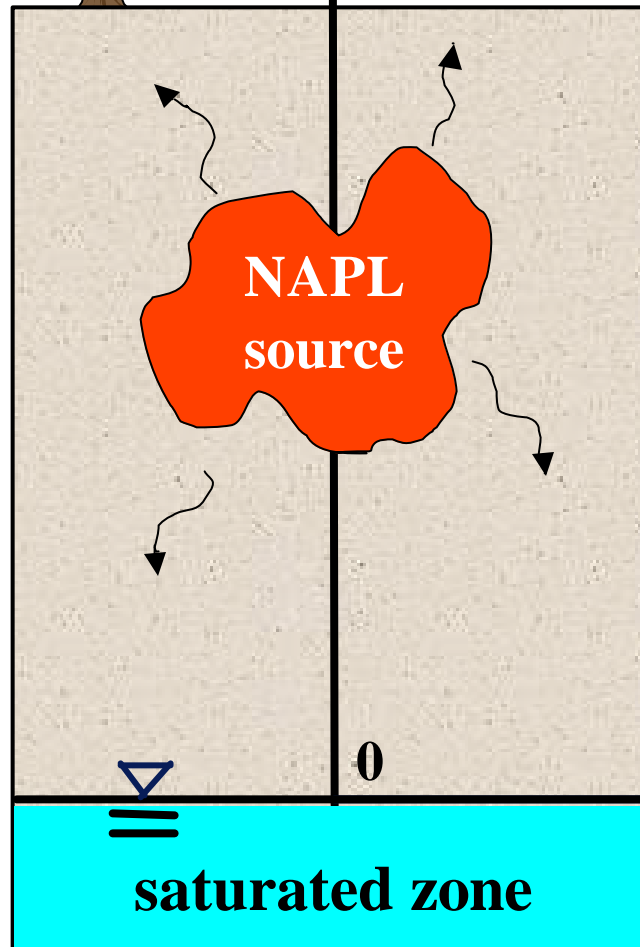
BOUNDARY CONDITIONS



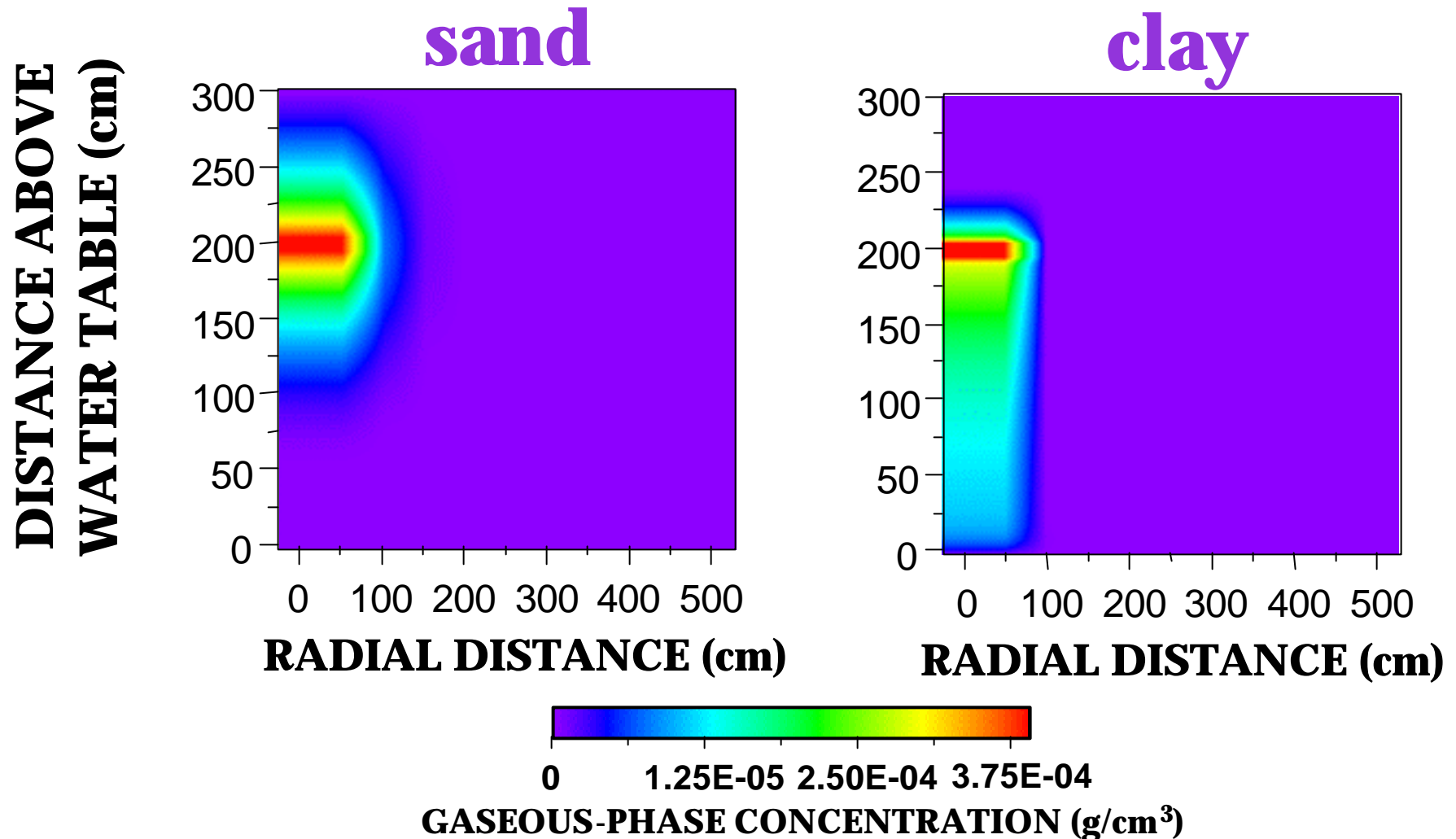
LAND SURFACE - atmospheric

SOURCE - constant concentration
- time dependent

WATER TABLE - MTBE, C_6H_6 : $C = 0$
 O_2 : impermeable



C_6H_6 DISTRIBUTION: CHRONIC RELEASE (infiltration rate = 20 cm/yr)



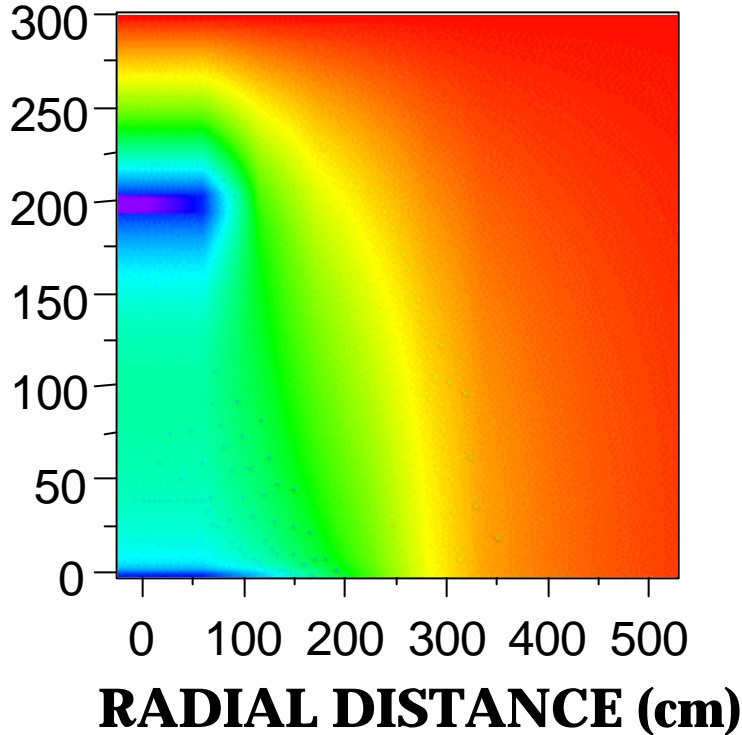
Lahvis and Rehmann, 2000

O₂ DISTRIBUTION: CHRONIC RELEASE

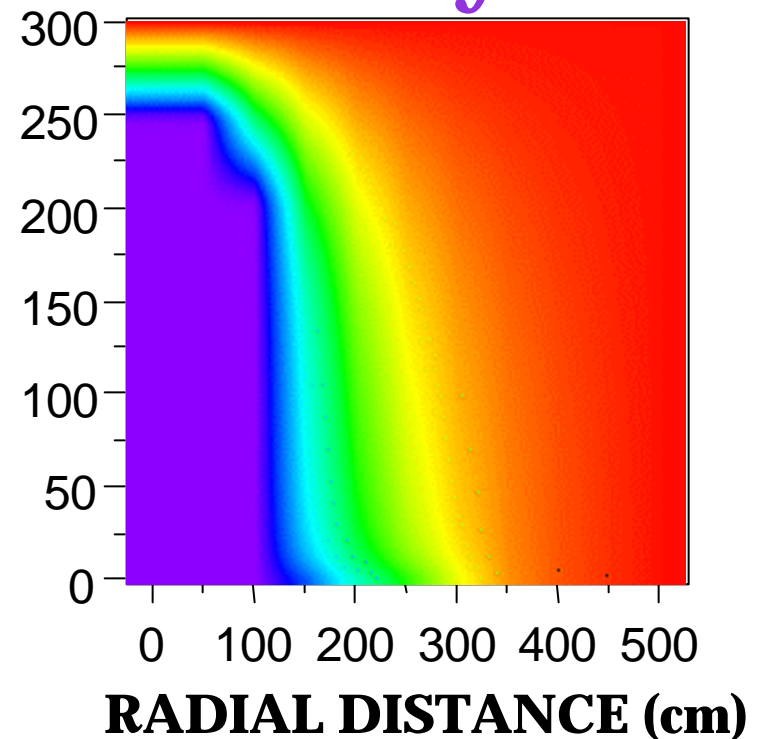
(infiltration rate = 20 cm/yr)

DISTANCE ABOVE
WATER TABLE (cm)

sand



clay

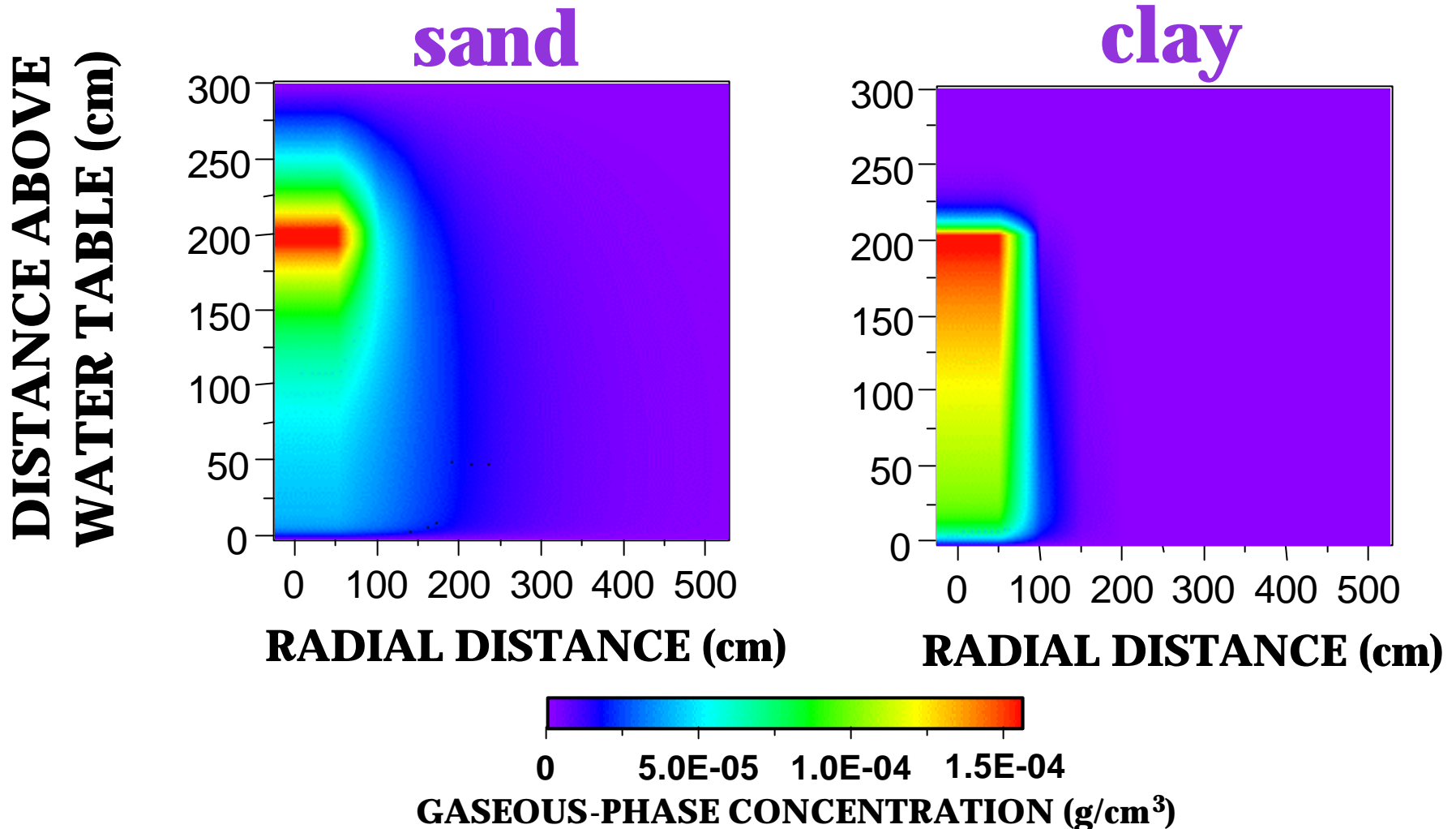


0 1.0E-04 2.0E-04
GASEOUS-PHASE CONCENTRATION (g/cm³)

Lahvis and Rehmann, 2000

MTBE DISTRIBUTION: CHRONIC RELEASE

(infiltration rate = 20 cm/yr)



WHAT WE LEARNED

- **enrichment in MTBE relative to benzene**
 - potential for ppm concentrations (chronic)
- **capillary zone limits mass transport**
- **diffusion to the atmosphere - significant mass-loss pathway**
- **breakthrough times variable**



Highlights:

Gasoline Release w/ MTBE

- MTBE preferentially released as vapor from UST
- MTBE:Benzene > 10 in vapor
- “soils-only” case also leads to preferential release of MTBE vapor
- MTBE Vapor Can Move Quickly, > 10 ft/yr
- Vadose zone recharge and biodegradation ‘enrich’ water phase relative to MTBE
- If BTEX and MTBE in groundwater, expect PSH (also consider if no O₂)

Local Examples

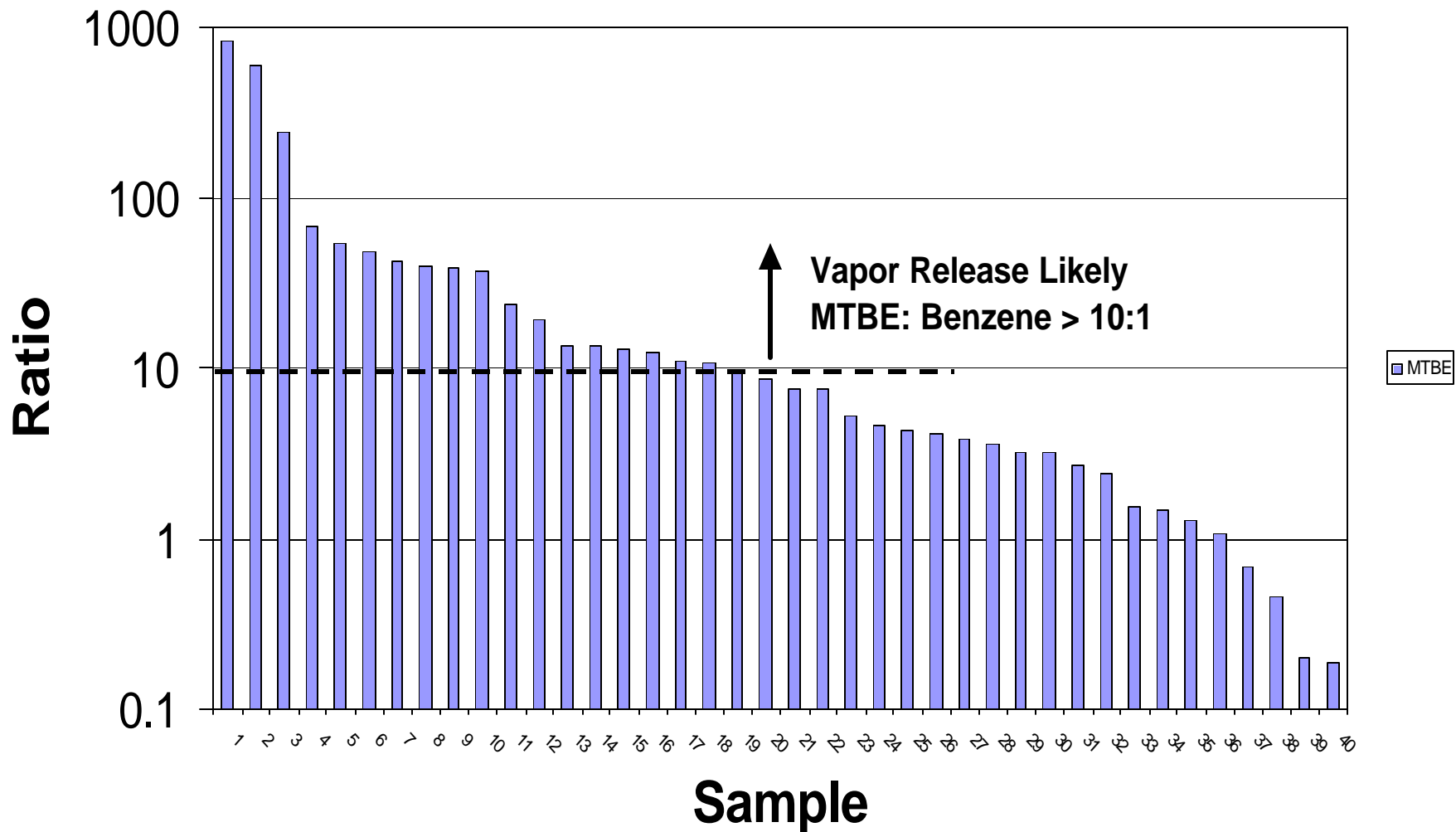
courtesy of Jim Schuck, DEH

- 12 active gas stations w/ “H numbers”
- 1997 soil vapor data at time of transaction
- Soil vapor samples at dispensers, lines, and USTs
- 92 vapor samples obtained

Soil Vapor Results

- All 12 sites w/detected MTBE
- 47% of probes encountered MTBE vapor
- range: 1480 to 357,000 ppb-v
- 72% encountered Benzene vapor
- range: 26 to 33,600 ppb-v
-DL for Benzene was 25 ppmv
-DL for MTBE was 1380 ppmv

MTBE: Benzene in Vapor

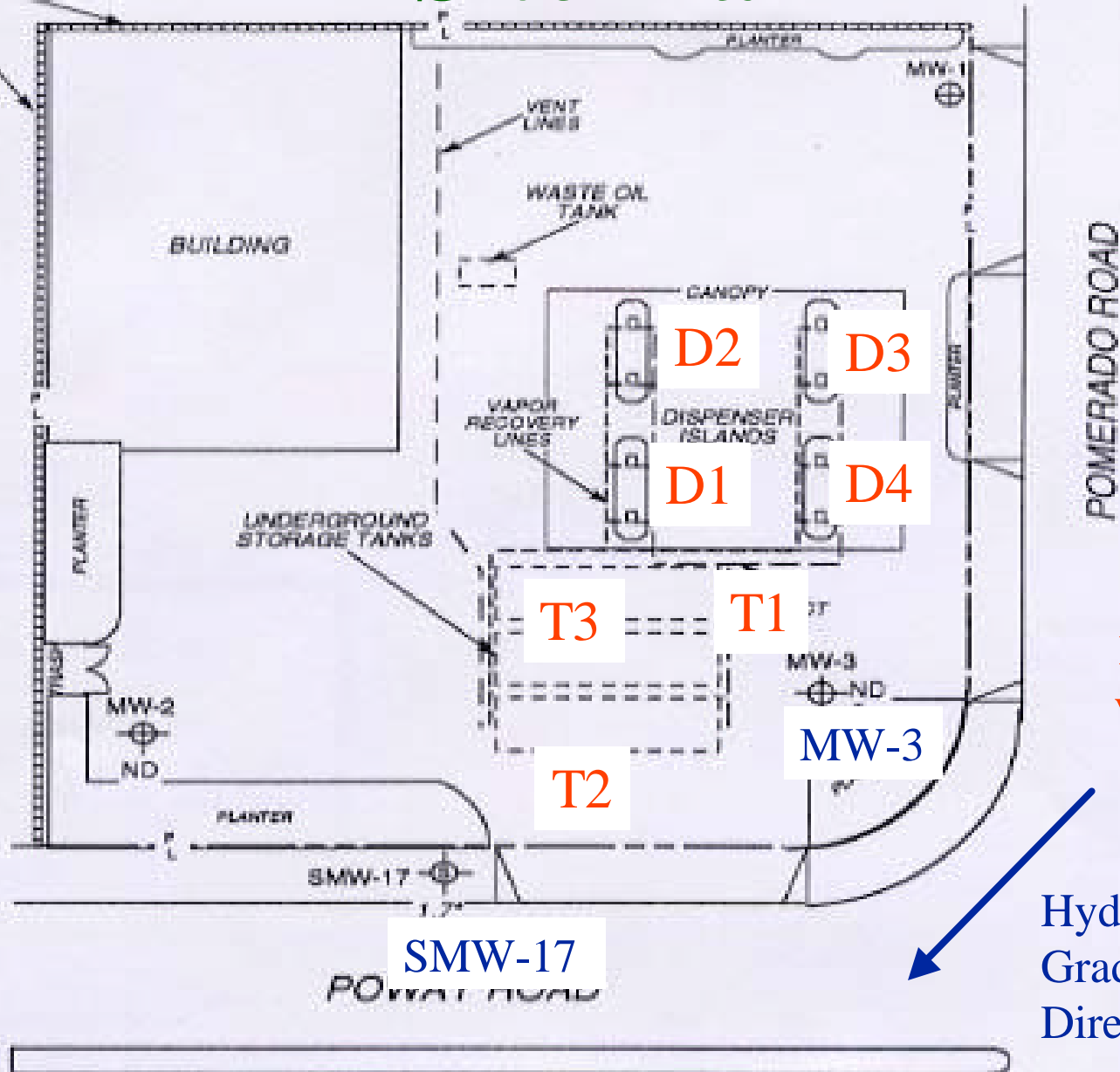


Case Example: “4887”

- 9/97 Soil Vapor Survey
- Work was done in 1998 to upgrade USTs
- Groundwater at 14 to 17 ft bgs
- Gravelly sand and Silty sand overlying DG

Site Plan

CONCRETE
BLOCK WALL



1997 soil
vapor points

Hydraulic
Gradient
Direction

Soil Vapor Data: 1997

Location	Benzene ppb-v	MTBE ppb-v	Ratio MTBE:B
T1	43	35,600	828
T2	117	ND<1380	...
T3	2290	30,300	13
D1	104	4010	39
D2	152	3570	24
D3	80	ND<1380	...
D4	90	ND<1380	...

Groundwater Data: 1999/2000

MW-3

	BTEX, ug/L	MTBE, ug/L
5/4/99	nd/nd/nd/nd	79,000
1/11/00	nd/nd/nd/nd	150,000
8/15/00	nd/nd/nd/nd	180,000

SMW-17

	BTEX, ug/L	MTBE, ug/L
5/4/99	46/2/1.7/4.6	30,000
1/11/00	nd/nd/nd/nd	29,000

Summary

- Yes, MTBE-only groundwater plumes can occur
- Source can be either liquid or vapor: both have similar behavior in vadose zone
- Expect MTBE:Benzene vapor > 10:1
- If soils are relatively dry, MTBE vapor will persist...and diffuse from release point
- Suggests SVE remedy may be feasible

The Future: Ethanol

